

STABILITY STUDY OF A 10% SODIUM BENZOATE ORAL SOLUTION

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BACKGROUND

Defects in the **urea cycle** are genetic diseases in which ammonia is accumulated. **Sodium benzoate** (SB) is conjugated with glycine giving rise to hippurate, excreted in the urine. Currently, no oral sodium benzoate preparation is commercialized.

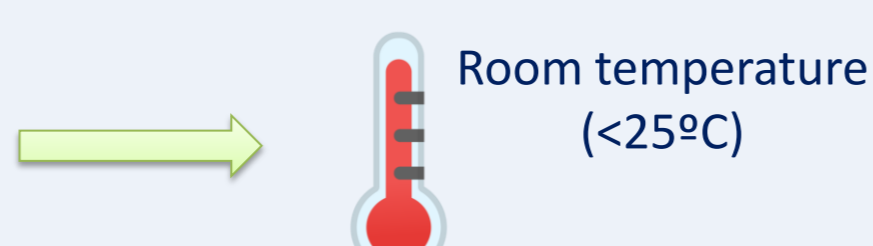
OBJECTIVE

To evaluate the stability of an **oral solution of 10% sodium benzoate** at two different storage conditions for the treatment of urea cycle disorders.

METHODS



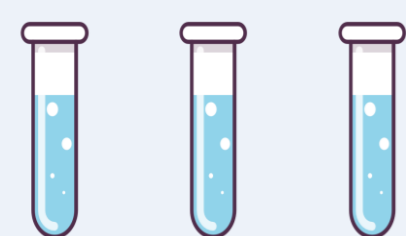
10% Sodium benzoate in water



Room temperature (<25°C)



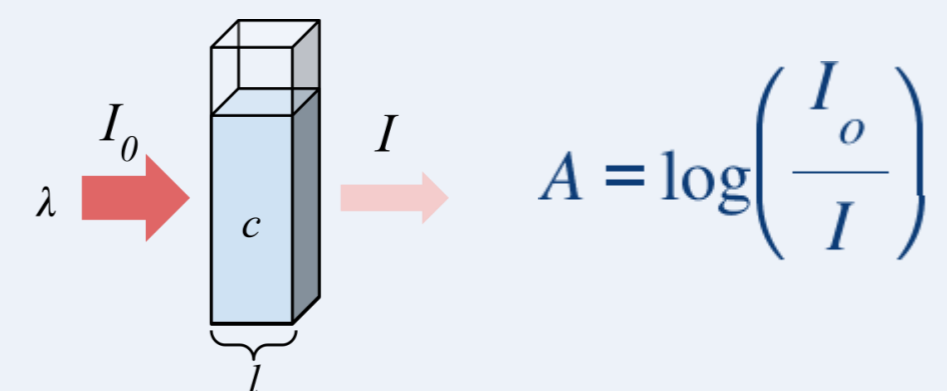
All formulations were stored protected from **light** during **30 days**.



Refrigerated 2-8 °C

Measurements

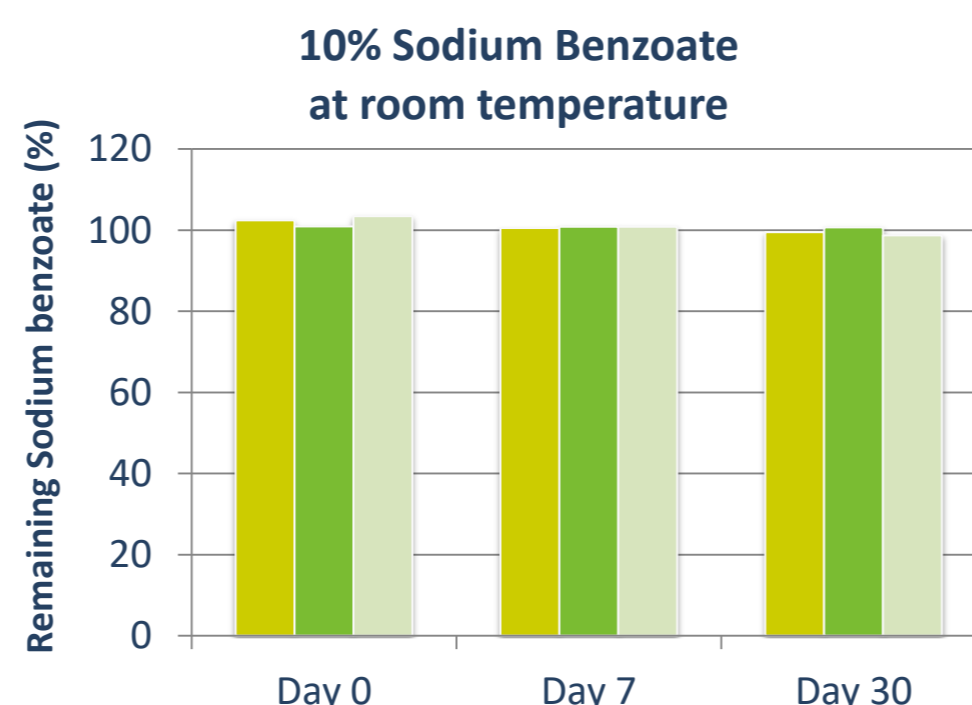
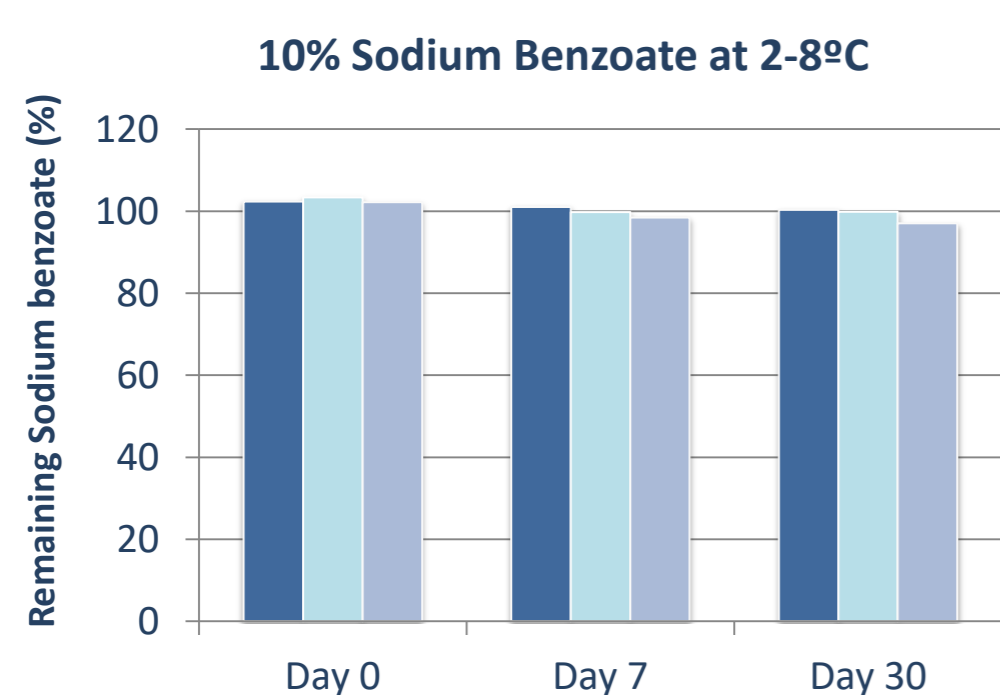
- ✓ Colour
- ✓ Opacity
- ✓ Precipitation
- ✓ pH
- ✓ Absorbance



$$A = \log\left(\frac{I_0}{I}\right)$$

All absorbance measurements were obtained with a Shimadzu® spectrophotometer model UVmini-1240 UV-Vis at **223 nm**.

RESULTS



Average degradation (%)		
Day	Refrigerated (2-8 °C)	Room temperature
7	2.82	1.49
30	3.48	2.55

After 30 days, no **colour** change, **opacity** or **precipitation** were observed. In all test solutions, **pH**-values remained unchanged.

CONCLUSION

The **10% Sodium benzoate oral solution**, used in urea cycle defects in pediatric patients, is **physically and chemically stable** when stored **protected from light** at **room temperature** or **refrigerated** (5°C±3°C) for at least **30 days**.

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